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SOLUTIONS MANUAL 3. Apply How many  
atoms of hydrogen-1 are in a 1.007-g  
sample? Recall that 1.007 amu is the

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mass of one atom of hydrogen-1. Round your answer to two significant digits.

1.007 g 1 H-1 atom  $\times$  1.672  $\times 10^{-24}$  g a.

6.023  $\times 10^{23}$  H-1 atoms 4.

## **The MoleThe Mole**

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14 SOLUTIONS MANUAL 32. Explain the

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similarities and differences between a 1 M solution of NaOH and a 1 m solution of NaOH. Both solutions contain NaOH (solute) dissolved in water (solvent). The 1 m solution contains 1 mole of NaOH per kilogram of water; the 1M solution 3  
\_ 5 \_ \_ \_ 1 \_ \_ \_ \_ \_ ( \_

## **Mixtures and Solutions**

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Solutions Manual Chemistry: Matter and Change • Chapter 9 145 CHAPTER 9 SOLUTIONS MANUAL 35. Aqueous solutions of potassium iodide and silver nitrate are mixed, forming the precipitate silver iodide. chemical equation:  $KI(aq) + AgNO_3(aq) \rightarrow KNO_3(aq) + AgI(s)$  complete ionic equation:  $K^+(aq) + I^-(aq) + Ag^+(aq) + NO_3^-(aq) \rightarrow K^+(aq) + NO_3^-(aq) + AgI(s)$

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(aq) AgI(s)

## **Chemical Reactions** **Chemical Reactions**

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Chapter 6 Solutions Manual CHAPTER 6  
SOLUTIONS MANUAL 4. Determine why  
producing 1 million fran-cium atoms per  
second is not enough to make

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measurements, such as density of boiling point. Even one million atoms collected together as a solid are microscopic. A grain of salt contains about  $10^{15}$  sodium atoms.

## **The Periodic Table and Periodic Law**

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Chapter 8 Solutions Manual 45.  $O_3$   $O_2$   
 $O_2$   $O_2$  46. Challenge Draw the Lewis  
resonance structure for the ion  $SO_3^{2-}$ .  
 $S$   $O_2^{2-}$   $O$   $O$   $O$   $S^{2-}$   $S^{2-}$  Draw the  
expanded octet Lewis structure of each  
molecule. 47.  $ClF_3$   $F$   $Cl$   $F$   $F$  48.  $PCl_5$   $Cl$   
 $Cl$   $Cl$   $Cl$   $Cl$   $P$  49. Challenge Draw the  
Lewis structure for the molecule formed  
when six fluorine atoms and



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## **Covalent Bonding**

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