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Physical Science Reading and Study Workbook Level B Chapter 6 55 IPLS Chapter 6 Chemical Bonds Summary 6.1 Ionic Bonding When the highest occupied energy level of an atom is filled with electrons, the atom is stable and not likely to react. • The chemical properties of an element depend on the number of valence electrons.

Chapter 6 Chemical Bonds - Henry County School District

CHAPTER 6 REVIEW Chemical Bonding SECTION 5 SHORT ANSWER Answer the following questions in the space provided.

1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms. Pairs of valence electrons repel one another. 2. In water, two hydrogen atoms are bonded to one oxygen atom. Why isn't water a linear molecule?

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6 Chemical Bonding

Writing Formulas for Ionic Compounds What is the ratio of the ions in magnesium iodide? What is the formula for magnesium iodide? 1. Read and Understand

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Chapter 6 Chemical Bonds... Physical Science Reading and Study Workbook Level B Chapter 6 65 IPLS Section 6.4 The Structure of Metals (pages 176-181) This section discusses metallic bonds and the properties of metals. It also explains how the properties of an alloy are controlled. <http://mrmicco.weebly.com/uploads/1/2/5/8/12584959/psgs1064b.pdf>

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